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Chapter 3 The Cellular Level of OrganizationChapter 9 Lecture Lesson 1 - Voltage, Current, Resistance (Engineering Circuit Analysis)

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1 a)The relative atomic mass of an element is the weighted average mass of the isotopes of the element. (1) It is measured on a scale on which a carbon-12 atom has a mass of exactly 12 units.(1). b)The total mass of 100 atoms = $(75 \times 35) + (25 \times 37) = 3550$ (1) The average mass of 1 atom = $3550/100 = 35.5$ (1)

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Chemistry 12th Edition Chang, Raymond; Goldsby, Kenneth Publisher McGraw-Hill Education ISBN 978-0-07802-151-0

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